

Thermodynamics of Energy Conversion and Storage

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EXERCISES 13

1) Calculate the theoretical COP for a heat pump in Switzerland if the water temperature for heating in the house is 50°C and the outside temperature is 10°C, 0°C and -10°C.

$$\begin{aligned} \text{COP} = T_1/\Delta T &= 323.15\text{K}/40\text{K} = 8 \\ &= 323.15\text{K}/50\text{K} = 6.4 \\ &= 323.15\text{K}/60\text{K} = 5.4 \end{aligned}$$

2) Calculate the heating power for a building with a heat pump as a function of temperature and surface area.

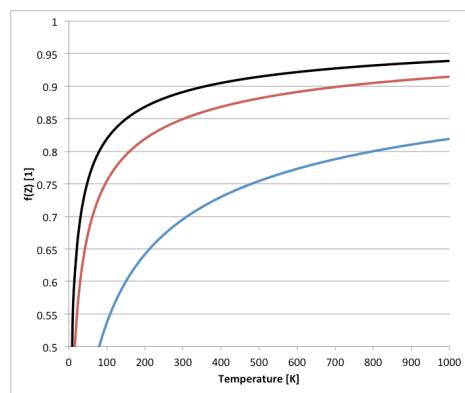
$$\begin{aligned} P_{\text{loss}} &= A \cdot \Delta T \cdot 1 \text{ W} \cdot \text{m}^{-2} \cdot \text{K}^{-1} \\ \text{COP} &= T_1/\Delta T \\ P_{\text{loss}}/\text{COP} &= P_{\text{el}} = A \cdot \Delta T^2 \cdot T_1^{-1} \cdot 1 \text{ W} \cdot \text{m}^{-2} \cdot \text{K}^{-1} \end{aligned}$$

3) Calculate the efficiency of a thermoelectric element with Z-T = 1 working between 100°C (1000°C) and 20°C. .

$$\begin{aligned} \eta_{\text{max}} &= \frac{P}{\dot{Q}_h} = \frac{\Delta T}{T_h} \cdot \frac{\sqrt{1+Z \cdot T} - 1}{\sqrt{1+Z \cdot T} + 1} \\ \eta &= \Delta T \cdot T_1^{-1} \cdot (\sqrt{1+373} - 1) / (\sqrt{1+373} + 1) = 0.90 \cdot 0.21 = 19\% \\ \eta &= \Delta T \cdot T_1^{-1} \cdot (\sqrt{1+1273} - 1) / (\sqrt{1+1273} + 1) = 0.97 \cdot 0.76 = 75\% \end{aligned}$$

4) Calculate the effect of increasing ZT on the efficiency.

$$F(Z) = \frac{\sqrt{1+Z \cdot T} - 1}{\sqrt{1+Z \cdot T} + 1} \quad Z = 0.1 \text{ blue, } Z = 0.5 \text{ red, } Z = 1 \text{ black}$$



5) Explain the function of cooling power of a Peltier element vs. the current. .

With increasing current the heat transported increases linearly and the Joules heat produced increases. Therefore, the cooling power goes through a maximum as a function of the current.